

Multiple Choice Answers

Q25: A

Multiple Choice Answers

Q24: B

Multiple Choice Answers

Q22: C

Multiple Choice Answers

Q23: D

Q24: B

Multiple Choice Answers

Q26: A

3(a)	VII
3(b)	6
3(c)(i)	tennessine / Ts / Ts ₂
3(c)(ii)	fluorine / F / F ₂
3(d)	halide(s)
3(e)	4
3(f)	krypton
3(g)(i)	$Cl_2 + 2I^- \rightarrow 2Cl^- + I_2$ M1 Cl ⁻ M2 correct ionic equation
3(g)(ii)	M1 I ⁻ loses e ⁻ M2 Cl ₂ gains e ⁻
3(h)	M1 grey-black M2 solid