

4.2 Hydrogen-oxygen fuel cells

Name: _____ Class: _____ Date: _____

Total: 7 marks

Objective

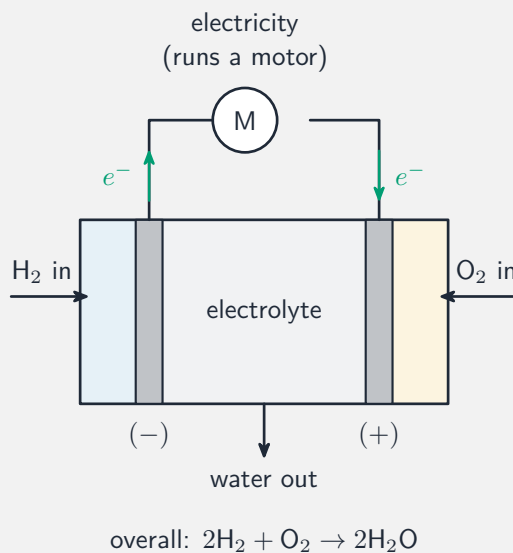
Build the skills to answer exam questions on **hydrogen–oxygen fuel cells** 氢氧燃料电池—how they work and how they compare with a petrol engine.

You must be able to:

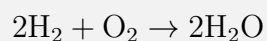
- give the overall reaction in a hydrogen–oxygen fuel cell
- state the advantages and disadvantages vs a petrol engine
- explain why the only product is water

1 Worked examples

■ The fuel cell



A fuel cell turns chemical energy straight into electricity; water is the only product



- **advantage:** only product is water —no air pollution.
- **disadvantage:** hydrogen is hard/dangerous to store; making it may use fossil fuels.

2 Practice

2.1 Write the overall equation for a hydrogen–oxygen fuel cell. [1]

2.2 State the only chemical product of the fuel cell. [1]

3 Exam-style questions

3.1 A key advantage of a hydrogen fuel cell over a petrol engine is that it: [1]

- **A** is easier to refuel
 - **B** produces only water, not pollutants
 - **C** uses no energy
 - **D** stores its fuel more easily
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3.2 Compare a hydrogen–oxygen fuel cell with a petrol engine.

(a) Give one advantage of the fuel cell. [1]

(b) Give one disadvantage of the fuel cell. [1]

3.3 Explain why using a hydrogen fuel cell may still cause some pollution overall. [2]

4 Go further

You are now ready for the real exam questions on this subtopic. Open the **4.2 Hydrogen–oxygen fuel cells** past-paper sheet in the Library, or try these in **Practice mode**:

- 0620/22 N25 —Q12 (fuel cells)
- 0620/23 N25 —Q15 (fuel cells)

Solutions

2.1 $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$.

2.2 water.

3.1 B. It produces only water, not pollutants.

3.2 (a) the only product is water, so it does not pollute the air.

(b) hydrogen is hard/dangerous to store and transport.

3.3 the hydrogen fuel is often made using energy from fossil fuels; burning those fuels releases pollutants/carbon dioxide elsewhere.