

Revision Sheet

IGCSE Chemistry · Unit 6 Chemical reactions

11 vocabulary items · 35 marks of exam questions (13 questions)

Part 1 · Vocabulary review answer key. Section A wants the exact English spelling.

Question	Answer	Marks
1	rate of reaction	1
2	forward reaction	1
3	collision theory	1
4	catalyst	1
5	equilibrium	1
6	Contact process	1
7	iron	1
8	sulfur	1
9	oxidation number	1
10	oxidising agent	1
11	reducing agent	1

Multiple Choice Answers

Q12: A

Q13: D

Q14: B

Q15: D

Multiple Choice Answers

Q16: D

Q17: C

Q18: C

Q19: C

Multiple Choice Answers

Q20: C

Q21: D

Q22: A

23(a)	<i>test:</i> relights AND <i>observations:</i> a glowing splint
23(b)(i)	lower gradient (at t2)
23(b)(ii)	M1 concentration (of H ₂ O ₂ particles) decreases (1) M2 frequency of collisions between particles decreases (1)
23(b)(iii)	M1 steeper curve which does not cross original curve and levels off before the original curve (1) M2 finishes at same volume (1)
23(c)	M1 mol O ₂ = $72 / 24000 = 0.003(00)$ (1) M2 mol H ₂ O ₂ = $M1 \times 2 = 0.002 \times 2 = 0.006(00)$ (1) M3 conc H ₂ O ₂ = $M2 \times 1000 / 20 = 0.006 \times 1000 / 20 = 0.3(00)$ (1) M4 M _r H ₂ O ₂ = 34 (1) M5 = $M4 \times M3 = 34 \times 0.3 = 10.2 \text{ (g / dm}^3\text{)}$ (1)
23(d)	any transition metal oxide

24(a)	M1 the rate of forward reaction equals (the rate of the) reverse reaction (1) M2 the concentrations of reactants and products are no longer changing (1)
24(b)(i)	M1 concentration decreases (1) M2 position of equilibrium shifts to left AND fewer moles of gas on left hand side (1)
24(b)(ii)	M1 concentration increases (1) M2 position of equilibrium shifts to the right AND forward reaction is endothermic (1)
24(b)(iii)	M1 no change (1) M2 catalysts do not affect position of equilibrium (1)
24(c)(i)	carbon dioxide
24(c)(ii)	M1 greenhouse gases absorb thermal energy (from the Earth) (1) M2 and M3 one mark each for any two of: <ul style="list-style-type: none">• energy (from the sun) absorbed by Earth's surface• Earth emits or reflects thermal energy• (greenhouse gases) reduces or stops thermal energy loss (into space)• increasing in amount of greenhouse gas results in a higher atmospheric temperature