

18.2 Fission (full E.4)

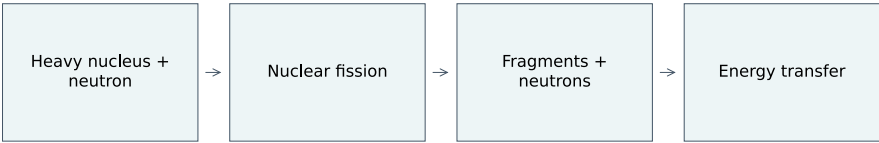
Name: _____ Class: _____ Date: _____ **Total: 21 marks**

KEY WORDS chain reaction • moderator • control rods
• nuclear waste

Objective

Practise the full guide E.4 toolkit: energy release in spontaneous and neutron-induced fission with conserved equations, chain reactions, the roles of moderator, control rods, heat exchanger and shielding in a power plant, the properties and management of fission products, and energy calculations from mass difference and binding energy. Original practice; not an official IB paper.

1 Key knowledge and worked examples



■ The E.4 rule set

$$E = \Delta mc^2, \quad 1\text{ u} = 931.5\text{ MeV}/c^2$$

Neutron-induced fission of U-235 gives Ba-141, Kr-92 and three neutrons – A and Z conserved exactly. The chain reaction is sustained when at least one product neutron triggers another fission.

■ Worked example: from mass defect to megajoules

A fission releases $\Delta m = 0.19\text{ u}$: $E = 0.19 \times 931.5 = 177\text{ MeV} \approx 2.8 \times 10^{-11}\text{ J}$. One kilogram of U-235 holds $(1000/235) \times 6.0 \times 10^{23} = 2.6 \times 10^{24}$ nuclei, so the total is about $7 \times 10^{13}\text{ J}$. Two habits: check A and Z on both sides before trusting a fission equation, and convert through $\text{u} \rightarrow \text{MeV} \rightarrow \text{J}$ in one clean chain.

2 Practice

2.1 Distinguish spontaneous fission from neutron-induced fission. [2]

2.2 State the role of the moderator in a nuclear reactor. [2]

2.3 State two properties of fission products that make them hazardous. [2]

2.4 State the function of control rods. [1]

3 Exam-style questions

3.1 Complete and balance: ${}^{235}_{92}\text{U} + {}^1_0\text{n} \rightarrow {}^{141}_{56}\text{Ba} + {}^{92}_{36}\text{Kr} + \dots$ State how many neutrons are released, and explain how this number makes a chain reaction possible. [3]

3.2 A fission reaction has a mass difference of $0.21\ u$. Calculate the energy released in MeV and in joules. [3]

3.3 Explain why fission fragments are radioactive even though the parent uranium is only weakly so, and describe one long-term management option for them. [4]

3.4 A reactor runs at a thermal power of $2.0 \times 10^9\ \text{W}$. Using $E = 2.8 \times 10^{-11}\ \text{J}$ per fission, calculate the number of fissions per second, and the mass of U-235 consumed per day if every nucleus fissions ($235\ u = 3.9 \times 10^{-25}\ \text{kg}$). [4]
