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Energy

(4.1.1.1)

- ### 1.2 Changes in energy – kinetic, elastic and gravitational (4.1.1.2)

- ### 1.3 Energy changes in systems – specific heat capacity (4.1.1.3, RP1)

- #### 1.4 Power (4.1.1.4) (4.1.1.4)

- (4.1.2.1, RP2)

- ### 1.6 Efficiency (4.1.2.2) (4.1.2.2)

- ### 1.7 National and global energy resources (4.1.3) (4.1.3)

- ## 2. Electricity

(4.2.1.1–4.2.1.2)

- ### 2.2 Current, resistance and potential difference (4.2.1.3, RP3)

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- ☐ Recall that the greater the resistance, the smaller the current for a given potential difference.
- ☐ Required practical 3: investigate how the resistance of a wire depends on its length at constant temperature, including meter placement, $R = V/I$, the proportional graph, zero error and keeping the wire cool.

2.3 Resistors and I–V characteristics (4.2.1.4, RP4) I-V (4.2.1.4, RP4)

- ☐ Explain that for some resistors resistance stays constant while in others it changes as the current changes.
- ☐ Describe the I-V graph of an ohmic conductor at constant temperature, a filament lamp and a diode.
- ☐ Explain the filament lamp graph: current heats the filament, resistance increases.
- ☐ State that thermistor resistance decreases as temperature increases and give the thermostat application.
- ☐ State that LDR resistance decreases as light intensity increases and give the switching-on-lights application.
- ☐ Required practical 4: investigate the I-V characteristics of circuit elements, including varying the potential difference, reversing the supply, and protecting the diode.

2.4 Series and parallel circuits (4.2.2) (4.2.2)

- ☐ For series components: state that the current is the same, the supply potential difference is shared, and the total resistance is the sum of the resistances.
- ☐ For parallel components: state that the potential difference is the same across each, the total current is the sum of the branch currents, and the total resistance of two resistors is less than the smallest individual resistance.
- ☐ Explain qualitatively why adding resistors in series increases total resistance while adding resistors in parallel decreases it.
- ☐ Calculate currents, potential differences and resistances in dc series circuits, using equivalent resistance.

2.5 Domestic uses and safety (4.2.3) (4.2.3)

- ☐ State that mains electricity is an ac supply with frequency 50 Hz and potential difference about 230 V in the UK.
- ☐ Explain the difference between direct and alternating potential difference.
- ☐ Identify the live, neutral and earth wires by insulation colour and state the job of each.
- ☐ Explain why a live wire may be dangerous even when a switch in the mains circuit is open.
- ☐ Explain the dangers of providing any connection between the live wire and earth.

2.6 Power and energy transfers in appliances (4.2.4.1–4.2.4.2)

(4.2.4.1–4.2.4.2)

- ☐ Use power = potential difference x current ($P = VI$) and power = current squared x resistance ($P = I^2 R$).
- ☐ Explain how the power transfer in a device relates to the potential difference across it, the current through it, and the energy transferred over time.
- ☐ Use energy transferred = power x time ($E = Pt$) and energy transferred = charge flow x potential difference ($E = QV$), with time in seconds.
- ☐ Describe how domestic appliances transfer energy to kinetic energy, heating or light, and relate power ratings to changes in stored energy in use.

2.7 The National Grid (4.2.4.3) (4.2.4.3)

- ☐ Describe the National Grid as a system of cables and transformers linking power stations to consumers.
- ☐ State that step-up transformers increase the transmission potential difference and step-down transformers decrease it for domestic use.
- ☐ Explain why the National Grid is an efficient way to transfer energy, using $P = VI$ and the cable power loss $P = I^2 R$.

2.8 Static electricity – physics only (4.2.5) — (4.2.5)

- ☐ Explain that rubbing insulating materials transfers electrons, leaving equal and opposite charges.
- ☐ Describe the forces between charged objects: like charges repel, unlike charges attract, as a non-contact force.
- ☐ Describe the production of static electricity and sparking by rubbing surfaces.
- ☐ Draw the electric field pattern for an isolated charged sphere.
- ☐ Explain the concept of an electric field and how it explains the non-contact force between charges and sparking.

3. Particle model of matter

3.1 Density of materials (4.3.1.1, RP5) (4.3.1.1, RP5)

- ☐ Use density = mass / volume with the units kg/m³ and g/cm³, converting between them.
- ☐ Use the particle model to explain the different states of matter and the differences in density between them.
- ☐ Recognise and draw simple diagrams that model solids, liquids and gases.
- ☐ Required practical 5: determine the densities of regular and irregular solid objects and liquids, using dimensions, a balance and a displacement technique.

3.2 Changes of state and internal energy (4.3.1.2–4.3.2.1) (4.3.1.2–4.3.2.1)

- ☐ Describe melting, freezing, boiling, evaporating, condensing and sublimating, and state that mass is conserved.
- ☐ Explain that changes of state are physical changes which recover the original properties when reversed.
- ☐ Define internal energy as the total kinetic and potential energy of all the particles in a system.
- ☐ Explain that heating either raises the temperature or produces a change of state.

3.3 Specific heat capacity and temperature changes (4.3.2.2) (4.3.2.2)

- ☐ Use $dE = m c d(\theta)$ for temperature changes, with the value of c interpreted per kilogram per degree Celsius.
- ☐ Interpret the specific heat capacity in particle terms.
- ☐ Solve for energy, mass, specific heat capacity or temperature change with unit conversions.

3.4 Specific latent heat (4.3.2.3) (4.3.2.3)

- ☐ Use energy for a change of state = mass x specific latent heat ($E = mL$).
- ☐ Define specific latent heat and distinguish fusion from vaporisation.
- ☐ Interpret heating and cooling graphs that include changes of state.

- ☐ ☐ Distinguish specific heat capacity from specific latent heat.

3.5 Particle motion in gases (4.3.3.1) (4.3.3.1)

- ☐ ☐ Describe gas molecules as in constant random motion.
- ☐ ☐ Relate the temperature of a gas to the average kinetic energy of its molecules.
- ☐ ☐ Explain gas pressure in terms of molecular collisions with the container walls.
- ☐ ☐ Explain qualitatively how the pressure of a fixed volume of gas changes with temperature.

3.6 Pressure in gases and work done on a gas – physics only (4.3.3.2–4.3.3.3) — (4.3.3.2–4.3.3.3)

- ☐ ☐ Use pressure $\times$ volume = constant for a fixed mass of gas at constant temperature.
- ☐ ☐ Calculate the new pressure or volume when either changes.
- ☐ ☐ Use the particle model to explain how increasing the volume of a gas decreases its pressure.
- ☐ ☐ (HT only) Explain how doing work on a gas increases its internal energy and can raise its temperature, for example in a bicycle pump.

4. Atomic structure

4.1 The structure of an atom; mass number and isotopes (4.4.1.1–4.4.1.2) ; (4.4.1.1–4.4.1.2)

- ☐ ☐ Describe the structure of the atom as a positive nucleus of protons and neutrons surrounded by electrons at different energy levels.
- ☐ ☐ Recall the order of magnitude of the atom's radius and that the nucleus is less than 1/10 000 of it, holding most of the mass.
- ☐ ☐ Use atomic number and mass number to find protons, neutrons and electrons.
- ☐ ☐ Define isotopes as atoms of the same element with different neutrons, and explain positive ions as atoms that have lost outer electrons.

4.2 Development of the model of the atom (4.4.1.3) (4.4.1.3)

- ☐ ☐ Describe the sequence: indivisible spheres, plum pudding model, nuclear model, Bohr orbits, protons, neutrons.
- ☐ ☐ Explain why the alpha scattering evidence led to the nuclear model.
- ☐ ☐ Describe the difference between the plum pudding model and the nuclear model.

4.3 Radioactive decay and nuclear radiation (4.4.2.1) (4.4.2.1)

- ☐ ☐ Describe radioactive decay as a random process in which unstable nuclei give out radiation.
- ☐ ☐ Define activity (becquerel) and count-rate.
- ☐ ☐ State the nature of alpha, beta, gamma and neutron radiation, with penetration, range in air and ionising power.
- ☐ ☐ Apply the properties to choose the best source for a given use.

4.4 Nuclear equations, half-lives and random decay (4.4.2.2–4.4.2.3) , (4.4.2.2–4.4.2.3)

- ☐ ☐ Write balanced nuclear equations for single alpha and beta decay, balancing atomic and mass numbers.
- ☐ ☐ Define half-life as the time for the number of nuclei or the count rate to halve.

- ☐ ☐ Determine half-life from given information or a graph.
- ☐ ☐ (HT only) Calculate the net decline, expressed as a ratio, after a given number of half-lives.

4.5 Radioactive contamination and background radiation (4.4.2.4–4.4.3.1) (4.4.2.4–4.4.3.1)

- ☐ ☐ Define radioactive contamination and irradiation, and state that irradiated objects do not become radioactive.
- ☐ ☐ Compare the hazards of contamination and irradiation.
- ☐ ☐ Describe suitable precautions against hazards from radioactive sources.
- ☐ ☐ Explain the importance of publishing and peer-reviewing studies of radiation effects.

4.6 Uses of nuclear radiation; choosing half-lives (4.4.3.2–4.4.3.3) ; (4.4.3.2–4.4.3.3)

- ☐ ☐ Describe natural and man-made sources of background radiation.
- ☐ ☐ State that dose depends on occupation and location, and subtract background from measurements.
- ☐ ☐ Explain how hazards differ according to half-life.
- ☐ ☐ Describe and evaluate uses of nuclear radiation in medicine for exploration of internal organs and destruction of unwanted tissue.

4.7 Nuclear fission and fusion – physics only (4.4.4) — (4.4.4)

- ☐ ☐ Describe nuclear fission: a neutron absorbed by a large unstable nucleus, the products, and the released energy.
- ☐ ☐ Explain chain reactions and the difference between controlled (reactor) and uncontrolled (weapon) versions.
- ☐ ☐ Draw and interpret diagrams representing fission and chain reactions.
- ☐ ☐ Describe nuclear fusion as the joining of two light nuclei with mass converting to radiation energy.

5. Forces

5.1 Scalars, vectors, types of force and resultant forces (4.5.1.1–4.5.1.4) , , (4.5.1.1–4.5.1.4)

- ☐ ☐ Distinguish scalar and vector quantities, with examples of each.
- ☐ ☐ Represent vectors as arrows with length for magnitude.
- ☐ ☐ Classify contact and non-contact forces with examples.
- ☐ ☐ Use weight = mass $\times$ gravitational field strength, recall the centre of mass and the newtonmeter.
- ☐ ☐ Calculate the resultant of collinear forces; (HT) use free-body diagrams, resolve forces and find resultants by scale drawing.

5.2 Work done and energy transfer (4.5.2) (4.5.2)

- ☐ ☐ Use work done = force $\times$ distance moved along the line of action of the force.
- ☐ ☐ Recall 1 joule = 1 newton-metre and convert between them.
- ☐ ☐ Describe the energy transfer when work is done, including the temperature rise from work against friction.

5.3 Forces and elasticity (4.5.3, RP6) (4.5.3, RP6)

- ☐ ☐ ☐ Explain why more than one force is needed to stretch, bend or compress a stationary object.
- ☐ ☐ ☐ Distinguish elastic and inelastic deformation.
- ☐ ☐ ☐ Use force = spring constant x extension and $E = 0.5 k e^2$ below the limit of proportionality.
- ☐ ☐ ☐ Interpret force-extension data and graphs; calculate a spring constant as the gradient.
- ☐ ☐ ☐ Required practical 6: investigate the relationship between force and extension for a spring.

5.4 Moments, levers and gears – physics only (4.5.4) , – (4.5.4)

- ☐ ☐ ☐ Use moment of a force = force x perpendicular distance from the pivot.
- ☐ ☐ ☐ Apply the balance of clockwise and anticlockwise moments.
- ☐ ☐ ☐ Explain how levers and gears transmit the rotational effects of forces.

5.5 Pressure and pressure differences in fluids – physics only (4.5.5) – (4.5.5)

- ☐ ☐ ☐ Use pressure = force normal to a surface / area of the surface.
- ☐ ☐ ☐ (HT) Use pressure = height x density x g for a column of liquid.
- ☐ ☐ ☐ Explain upthrust and the factors for floating and sinking.
- ☐ ☐ ☐ Explain why atmospheric pressure decreases with height.

5.6 Describing motion along a line (4.5.6.1.1–4.5.6.1.5, RP7) (4.5.6.1.1–4.5.6.1.5, RP7)

- ☐ ☐ ☐ Distinguish distance from displacement and speed from velocity.
- ☐ ☐ ☐ Recall typical speeds for walking, running, cycling and sound in air.
- ☐ ☐ ☐ Use $s = vt$ and average speed; read distance-time graphs by gradient with (HT) tangents.
- ☐ ☐ ☐ Use $a = \text{change in velocity} / \text{time}$; velocity-time gradients and (HT) areas; $v^2 \text{ squared minus } u^2 \text{ squared} = 2as$.
- ☐ ☐ ☐ Describe motion in a fluid reaching terminal velocity.

5.7 Forces, accelerations and Newton's laws (4.5.6.2) , – (4.5.6.2)

- ☐ ☐ ☐ State and apply Newton's first law, including (HT) inertia.
- ☐ ☐ ☐ Use resultant force = mass x acceleration; (HT) inertial mass.
- ☐ ☐ ☐ State and apply Newton's third law to equilibrium situations.
- ☐ ☐ ☐ Required practical 7: investigate the effect of force on acceleration at constant mass, and mass at constant force.

5.8 Forces and braking (4.5.6.3) (4.5.6.3)

- ☐ ☐ ☐ Define stopping distance as thinking distance plus braking distance.
- ☐ ☐ ☐ Explain reaction-time factors; measure human reaction times.
- ☐ ☐ ☐ Explain how speed, road/weather and vehicle condition affect braking distance.
- ☐ ☐ ☐ Explain braking as frictional work on the kinetic store and the dangers of large decelerations.

5.9 Momentum – HT only (4.5.7) – (4.5.7)

- ☐ ☐ ☐ Use momentum = mass x velocity.
- ☐ ☐ ☐ Apply conservation of momentum to collisions in a closed system.
- ☐ ☐ ☐ Use force = change in momentum / time.

- ☐ ☐ ☐ Explain safety features by the longer impact time reducing the force.

6. Waves

6.1 Transverse and longitudinal waves; properties of waves (4.6.1.1–4.6.1.2, RP8) ; (4.6.1.1–4.6.1.2, RP8)

- ☐ ☐ ☐ Describe the difference between transverse and longitudinal waves with examples.
- ☐ ☐ ☐ Describe evidence that the wave, not the material, travels.
- ☐ ☐ ☐ Use amplitude, wavelength, frequency and period; apply period = 1/frequency and wave speed = frequency x wavelength.
- ☐ ☐ ☐ Describe methods to measure the speed of sound in air and of ripples on water.
- ☐ ☐ ☐ Required practical 8: measure frequency, wavelength and speed in a ripple tank and in a solid.
- ☐ ☐ ☐ (Physics only) Relate velocity, frequency and wavelength changes when sound passes between media.

6.2 Reflection, sound and waves for detection – physics only (4.6.1.3–4.6.1.5) , – (4.6.1.3–4.6.1.5)

- ☐ ☐ ☐ Construct ray diagrams for reflection at a surface; describe absorption and transmission at interfaces.
- ☐ ☐ ☐ Required practical 9: investigate reflection by different surfaces and refraction by different substances.
- ☐ ☐ ☐ (HT) Describe sound conversion to vibrations of solids, the ear, and the 20 Hz to 20 kHz human hearing range.
- ☐ ☐ ☐ (HT) Explain ultrasound imaging by partial reflection at boundaries, and seismic P-wave/S-wave exploration.

6.3 Electromagnetic waves (4.6.2.1–4.6.2.4) (4.6.2.1–4.6.2.4)

- ☐ ☐ ☐ Describe EM waves as transverse, forming a continuous spectrum, all at the same speed in vacuum or air.
- ☐ ☐ ☐ Recite the order of the spectrum from radio to gamma in wavelength and frequency.
- ☐ ☐ ☐ Give uses of each band and (HT) explain their suitability.
- ☐ ☐ ☐ State the hazards of ultraviolet, X-rays and gamma rays; interpret radiation dose data.
- ☐ ☐ ☐ (HT) Explain how substances absorb, transmit, refract or reflect EM waves differently with wavelength; construct refraction ray and wavefront diagrams.

6.4 Lenses and visible light – physics only (4.6.2.5–4.6.2.6) – (4.6.2.5–4.6.2.6)

- ☐ ☐ ☐ Construct ray diagrams for convex and concave lenses; distinguish real and virtual images.
- ☐ ☐ ☐ Use magnification = image height / object height as a unitless ratio.
- ☐ ☐ ☐ Explain colour by differential reflection and absorption; filters by transmission; specular vs diffuse reflection.

6.5 Black body radiation – physics only (4.6.3, RP9) – (4.6.3, RP9)

- ☐ ☐ ☐ State that all bodies emit and absorb infrared radiation, more when hotter.
- ☐ ☐ ☐ Define a perfect black body as complete absorber and best emitter.
- ☐ ☐ ☐ Relate intensity and wavelength distribution of emission to temperature.

- ☐☐☐ (HT) Explain constant temperature as balanced absorption and emission, and apply to the Earth's temperature factors.

7. Magnetism and electromagnetism

7.1 Permanent and induced magnetism, magnetic fields (4.7.1) (4.7.1)

- ☐☐☐ Describe attraction and repulsion between permanent magnet poles as a non-contact force.
- ☐☐☐ Distinguish permanent from induced magnets, and recall that induced magnetism always causes attraction.
- ☐☐☐ Describe the magnetic field and its direction; recall the four magnetic materials.
- ☐☐☐ Explain how a plotting compass shows field directions, and the compass evidence for the Earth's field.

7.2 Electromagnetism and the motor effect (4.7.2) (4.7.2)

- ☐☐☐ Describe the magnetic field around a current-carrying wire and the strong uniform field inside a solenoid; explain electromagnets.
- ☐☐☐ Draw the field patterns for a straight wire and a solenoid with directions.
- ☐☐☐ Apply Fleming's left-hand rule and $F = BIl$ to conductors at right angles to a field.
- ☐☐☐ Explain the rotation of a motor coil and the role of the split-ring commutator.
- ☐☐☐ (Physics only) Explain how loudspeakers and headphones convert current variations to sound pressure variations.

7.3 Induced potential, the generator effect and transformers – physics only, HT (4.7.3)

- ☐☐☐ State the conditions for the generator effect and the factors affecting the size and direction of the induced pd.
- ☐☐☐ Explain alternators (ac) and dynamos (dc) and interpret their pd-time graphs.
- ☐☐☐ Explain how moving-coil microphones convert sound to current variations.
- ☐☐☐ Use the transformer turns and power equations; explain induction between coils and the advantage of high-pd transmission.

8. Space physics – physics only (4.8) (4.8)

8.1 Our solar system and the life cycle of a star (4.8.1.1–4.8.1.2)

(4.8.1.1–4.8.1.2)

- ☐☐☐ Describe the solar system: one star, eight planets, dwarf planets and natural satellites; part of the Milky Way.
- ☐☐☐ Explain the Sun's formation from a nebula pulled together by gravity, and the fusion equilibrium of a main-sequence star.
- ☐☐☐ Describe the life cycles of a Sun-sized star and of a much more massive star.
- ☐☐☐ Explain how fusion processes produce the naturally occurring elements and how a supernova forms and distributes elements heavier than iron.

8.2 Orbital motion, natural and artificial satellites (4.8.1.3) (4.8.1.3)

- ☐☐☐ Describe gravity as the force maintaining circular orbits of planets and satellites.
- ☐☐☐ Describe the similarities and distinctions between planets, their moons and artificial satellites.

- ☐☐☐ (HT only) Explain qualitatively how circular orbits involve changing velocity but unchanged speed.

- ☐☐☐ (HT only) Explain how a stable orbit must change radius when the speed changes.

8.3 Red-shift – physics only (4.8.2) (4.8.2)

- ☐☐☐ Describe red-shift as an observed increase in wavelength of light from most distant galaxies.
- ☐☐☐ State the link between distance, recession speed and the size of the red-shift.
- ☐☐☐ Explain how red-shift is evidence for an expanding universe and the Big Bang theory.
- ☐☐☐ Describe how observations, including the 1998 supernova results, lead to theories, and name current unknowns such as dark matter and dark energy.