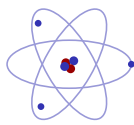


Particle Physics
AS Physics ■ Topic 11

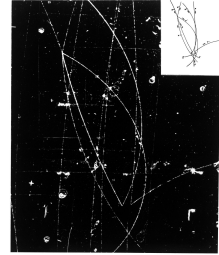
A-Level Physics



Particle Physics

What we will cover

- 1 α -scattering & the nucleus
- 2 Nuclide notation & isotopes
- 3 Radioactive emissions
- 4 Decay equations
- 5 Quarks & fundamental particles
- 6 Recap

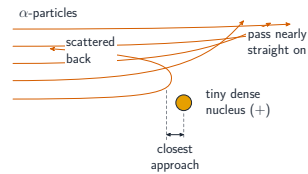


Production and decay of neutral lambda and anti-lambda hyperons

1 The nuclear atom

Particle Physics
The nuclear atom

α -particle scattering

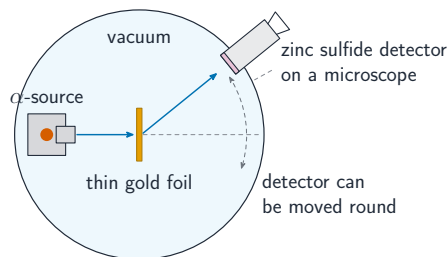


- mostly pass straight through, with very little **deflection** 偏转,
- sometimes deflect through small angles,
- rarely (about 1 in 8000) deflect through angles greater than 90° .
- the atom is **mostly empty space** (most α -particles pass straight through),

Particle Physics
The nuclear atom

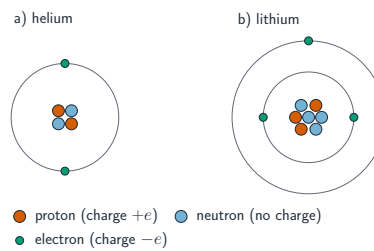
The experiment itself

Almost every α went straight through; a very few came straight back. That is only possible if the positive charge sits in a tiny, massive core.



Particle Physics
The nuclear atom

The nuclear model



- **protons** 质子 (+e) and **neutrons** 中子 (0) in the nucleus, $\approx 1 \text{ u}$ each
- **electrons** 电子 (-e) around it, $\frac{1}{1836}$ of the mass

Nuclide notation and isotopes

$\begin{matrix} A \\ Z \end{matrix} X$ nucleon number
proton number

- neutrons $N = A - Z$
- isotopes** 同位素: same Z , different A (e.g. ${}^{12}_6\text{C}$, ${}^{14}_6\text{C}$)

In every nuclear process, A and **charge** are conserved.

Reading the notation

$\begin{matrix} A \\ Z \end{matrix} X$ A = nucleon number (protons + neutrons)
 Z = proton number

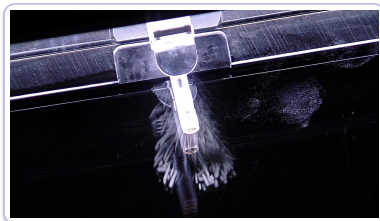
Top is the **nucleon number** 核子数 A , bottom the **proton number** 质子数 Z .
Isotopes 同位素 share Z and differ in A .

2 Radioactive emissions

Three kinds of radiation

α	β	γ
He nucleus $+2e$, $\approx 4\text{ u}$ stopped by paper discrete energy	fast electron/positron $\pm e$, tiny mass stopped by Al continuous energy	photon 光子 0 charge, 0 mass attenuated by lead discrete energy

Tracks in a cloud chamber

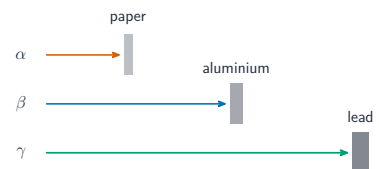


α tracks are short, straight and thick — heavily ionising, quickly stopped.

Penetrating power

- α — paper
- β — a few mm of aluminium
- γ — only attenuated by lead

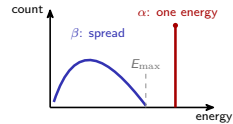
Ionising power is the reverse order: α strongest.



Antiparticles and the neutrino

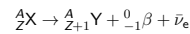
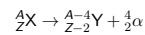
Every particle has an **antiparticle** 反粒子: same mass, opposite charge (positron = anti-electron). β -decay always emits a third particle:

- β^- : an **antineutrino** 反中微子 $\bar{\nu}_e$
- β^+ : a **neutrino** 中微子 ν_e



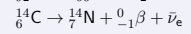
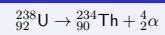
α : energy shared by 2 bodies $\rightarrow$ discrete. β : shared by 3 $\rightarrow$ continuous – the neutrino takes a varying share.

Writing decay equations



Balance A (top) and Z (bottom) on both sides.

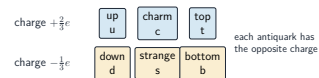
U-238 (α) and C-14 (β^-)



check: A $238 = 234 + 4$ ✓ Z $92 = 90 + 2$ ✓

3 Fundamental particles

Quarks



Six **quark** 夸克 flavours:

- u, c, t carry $+\frac{2}{3}e$
- d, s, b carry $-\frac{1}{3}e$

Each **antiquark** 反夸克 has the opposite charge.

Hadrons: baryons and mesons

Particles built from quarks are **hadrons** 强子:

baryons 重子

three quarks
proton (uud), neutron (udd)

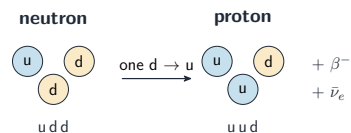
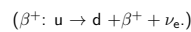
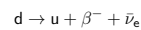
mesons 介子

one quark + one antiquark
e.g. $\pi^+ = u\bar{d}$

Protons and neutrons are **not** fundamental – they are baryons.

β -decay at the quark level

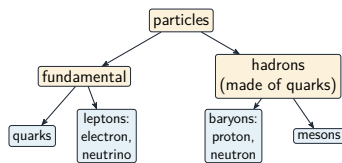
A neutron $\rightarrow$ proton: one **down** quark becomes an **up** quark.



Beta-minus decay: one down quark becomes an up quark, turning a neutron into a proton

Some particles are **fundamental** 基本粒子 (point-like, with no smaller parts as far as we know); others are built from fundamental ones.

Classifying particles



Fundamental: **quarks** 夸克 and **leptons** 轻子 (electron, neutrino).

Built from quarks: protons, neutrons, baryons, mesons.

4

Recap

Topic 11 – key takeaways

1 Nuclear atom

tiny dense + nucleus; A_ZX , A & charge conserved

2 Radiation

α (paper), β (Al), γ (lead)

3 β & neutrinos

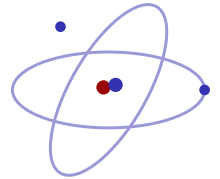
3 bodies $\rightarrow$ continuous spectrum

4 Quarks

hadrons from quarks; quarks & leptons fundamental

Check yourself

- 1 What did the rare large α deflections show?
- 2 ${}^{226}_{88}\text{Ra}$ decays by α – the daughter's A and Z ?
- 3 Which radiation is a photon?
- 4 Quark content of a neutron?



Answers 1. a tiny dense nucleus 2. $A = 222$, $Z = 86$ 3. γ 4. $u d d$