

Multiple Choice Answers

Q27: A

Multiple Choice Answers

Q27: B

Q28: A

Q29: D

Multiple Choice Answers

Q29: A

4(a)(i)	reduction
4(a)(ii)	$PCl_5 / PCl_3 / SOCl_2$ OR $KCl / NaCl$ AND conc. $H_2SO_4 /$ conc. H_3PO_4
4(a)(ii)	M1 $C_2H_5CH_2Cl + NaOH \rightarrow C_3H_6 + H_2O + NaCl$ M2 ethanol (solvent) AND heat / reflux
4(a)(iv)	conc. H_2SO_4 OR conc. H_3PO_4 OR Al_2O_3 AND heat
4(b)(i)	addition
4(b)(ii)	M1 (molecules that have the) same structural formula (and same molecular formula) M2 but different 3D / spatial arrangement of atoms / groups
4(b)(iii)	$\begin{array}{cccc} H & CH_3 & H & CH_3 \\ & & & \\ -C & -C & -C & -C- \\ & & & \\ H & H & H & H \end{array}$ M1 M2 optical isomerism M3 the methylated carbon has four different groups attached
4(b)(iv)	M1 non-biodegradability M2 harmful combustion products
4(c)(i)	2,3-dimethylbut-1-ene
4(c)(ii)	(The spectrum is of) X AND 41 is $[CH_2=CCH_3]^+ /$ 41 is $C_3H_5^+$ OR 43 is $[CH_3CHCH_3]^+ /$ 43 is $C_3H_7^+$ OR X would NOT have significant peak at 54 / $[(CH_3)_2C=C]^+ / C_4H_6^+$ OR X would NOT have significant peak at 42 due to $[(CH_3)_2C]^+ / C_3H_6^+$
4(c)(iii)	$([M+1]^+ / 34.7 \times 100 / 1.1 = 6)$ $[M+1]^+ = 2.29$