

38 Seven aldehydes and ketones

- contain only **one** oxygen atom per molecule
- have four or fewer carbon atoms in their structures.

Alkaline $I_2(aq)$ is added to each of these carbonyl compounds separately.

How many of these carbonyl compounds produce a yellow precipitate with alkaline $I_2(aq)$?

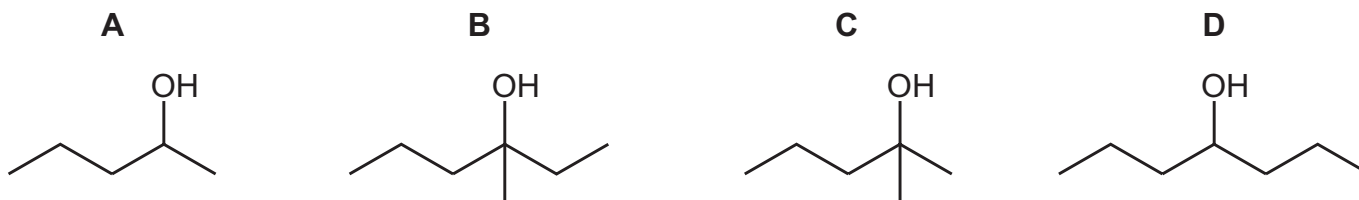
- A** 1 **B** 2 **C** 3 **D** 4

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34 An organometallic lithium compound, RLi , contains the nucleophile R^- .

CH_3CH_2Li reacts with pentan-2-one. The mechanism is nucleophilic addition. The first step produces an anion which is then protonated to form the final product.

Which organic product is formed?



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35 Butanone, $CH_3CH_2COCH_3$, and HCN mixed with a little KCN react together in a nucleophilic addition reaction.

Which description of the mechanism of this nucleophilic addition reaction is correct?

- A** The π bond pair of $C=O$ accepts H^+ from HCN and then CN^- acts as a nucleophile.
B A π bond pair from CN^- attacks the δ^+ C and HCN then donates H^+ to O.
C The lone pair on O accepts H^+ from HCN and then CN^- acts as a nucleophile.
D A lone pair from CN^- attacks the δ^+ C and then O^- accepts H^+ from HCN .