

36.1 Organic synthesis

Name: _____ Class: _____ Date: _____

Total: 11 marks

KEY WORDS reagent 试剂 • amine 胺 • carboxylic acid 羧酸 • amide 酰胺 • nitrile 腈

Objective

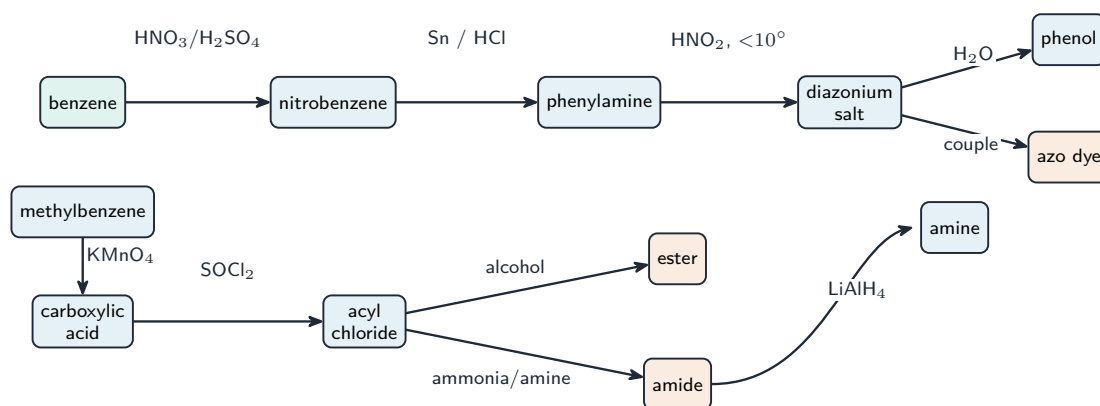
Build the skills to answer exam questions on **organic synthesis** 有机合成 — joining known reactions to build a target, working **backwards** from the product, and analysing a route.

You must be able to:

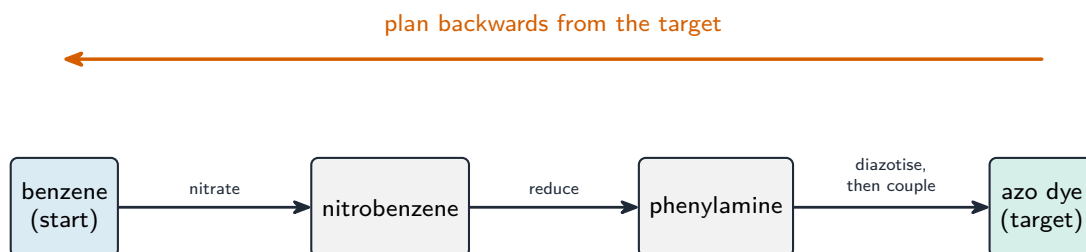
- choose reagents and conditions for each step of a synthesis
- devise a multi-step route by working backwards from the target
- state the reaction type at each step and identify likely by-products

1 Worked examples

■ Planning a route



*Two main chains: the aromatic chain (top) and the acyl-chloride chain (bottom).
Work backwards from your target*



Plan a synthesis by retrosynthesis

- **work backwards:** which single reaction makes the target, and from what? Repeat until you reach the start.
- **key steps:** benzene $\rightarrow$ (HNO₃/H₂SO₄) $\rightarrow$ nitrobenzene $\rightarrow$ (Sn/HCl, NaOH) $\rightarrow$ phenylamine $\rightarrow$ (HNO₂, <10 °C) $\rightarrow$ diazonium salt.
- carboxylic acid $\rightarrow$ (SOCl₂) $\rightarrow$ acyl chloride $\rightarrow$ (alcohol) $\rightarrow$ ester / $\rightarrow$ (amine) $\rightarrow$ amide.
- nitrile/amide $\rightarrow$ (LiAlH₄) $\rightarrow$ amine; halogenoalkane $\rightarrow$ (KCN) $\rightarrow$ nitrile (**adds one carbon**).

2 Practice

2.1 Give the reagents to convert benzene into nitrobenzene. [2]

2.2 State the reagent that reduces a nitrile to an amine. [1]

3 Exam-style questions

3.1 To lengthen a carbon chain by one carbon, a halogenoalkane is reacted with: [1]

A	B
C	D

A NH₃

B KCN

C LiAlH₄

D H₂O

3.2 In planning a synthesis, the best strategy is to:

[1]

A	B
C	D

A work forwards from any reagent

B work backwards from the target molecule

C guess reagents at random

D use only one reaction

3.3 Devise a synthesis of **phenylamine** from benzene, giving the reagents and conditions for each step.

[3]

3.4 A two-step synthesis converts bromoethane into propylamine via a nitrile.

(a) Give the reagents for each step.

[2]

(b) State **one** by-product/limitation of making an amine directly from a halogenoalkane and ammonia.

[1]
