

17.1 Aldehydes and ketones

Name: _____ Class: _____ Date: _____

Total: 13 marks

KEY WORDS nucleophilic addition 亲核加成 • tollens' reagent 托伦试剂 • carbonyl 羰基
 • aldehyde 醛 • ketone 酮

Objective

Build the skills to answer exam questions on **aldehydes and ketones** 醛和酮 — making them, **nucleophilic addition** 亲核加成 of HCN, and the tests that tell an aldehyde from a ketone.

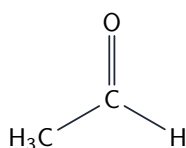
You must be able to:

- describe how aldehydes and ketones are made and reduced
- describe the nucleophilic addition mechanism of HCN
- use 2,4-DNPH, Fehling's and Tollens' tests, and the iodoform test

1 Worked examples

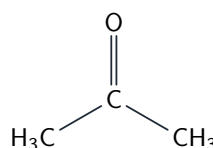
■ Carbonyl compounds

aldehyde



C=O carbon on the end
(also carries an H): -CHO

ketone

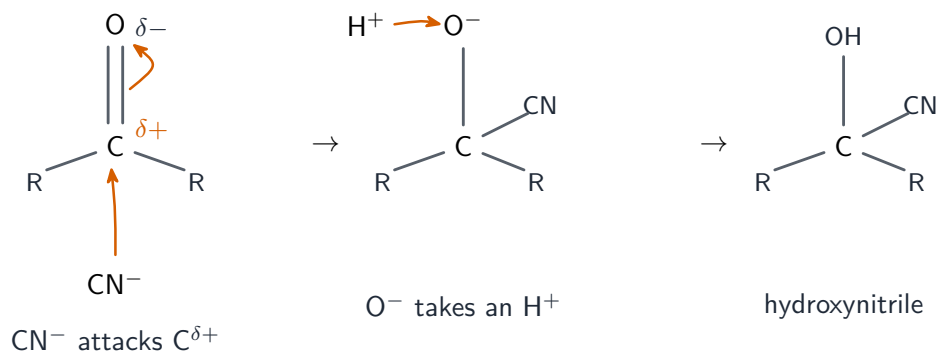


C=O carbon in the middle,
between two carbons

Both have the C=O carbonyl group: aldehyde = on the end (-CHO); ketone = in the middle

Made by oxidising alcohols: primary (distil) → **aldehyde**; secondary → **ketone**.
 Reduced back by NaBH₄/LiAlH₄. With HCN (KCN catalyst) → a **hydroxynitrile** (+1 carbon).

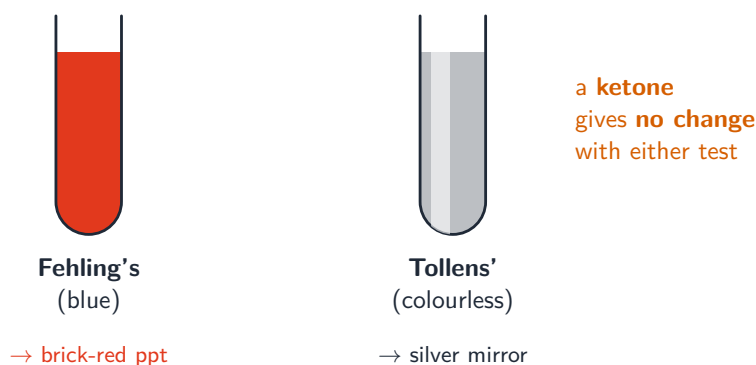
■ Nucleophilic addition of HCN



CN^- attacks the δ^+ carbon; the $\text{C}=\text{O}$ breaks to O^- ; O^- takes an H^+ to give the hydroxynitrile

■ Tests

tell an aldehyde from a ketone



2,4-DNPH (orange ppt) detects any carbonyl. Aldehyde: Fehling's → brick-red, Tollens' → silver mirror. Ketone: no change

The **iodoform test** (alkaline I_2) gives a yellow CHI_3 precipitate for a $\text{CH}_3\text{CO}-$ group.

2 Practice

2.1 State the difference in structure between an aldehyde and a ketone. [2]

2.2 Name the reagent used to detect the presence of any carbonyl compound. [1]

3 Exam-style questions

3.1 Which reagent gives a silver mirror with an aldehyde but no change with a ketone? [1]

A	B
C	D

- A 2,4-DNPH
- B Tollens' reagent
- C bromine water
- D NaBH₄

3.2 The reaction of HCN with a ketone is classified as: [1]

A	B
C	D

- A electrophilic addition
- B nucleophilic addition
- C free-radical substitution
- D elimination

3.3 (a) Describe the mechanism of the nucleophilic addition of HCN to ethanal. [3]

(b) State the reagent and condition to reduce propanone to an alcohol, and name the alcohol formed. [2]

3.4 Describe how you would use chemical tests to distinguish between propanal (an aldehyde) and propanone (a ketone). [3]